

Topical Symposium on Sustainable Surface Engineering Room Golden State Ballroom - Session TS1-ThP

Coatings for Batteries and Hydrogen Applications Poster Session

TS1-ThP-1 Hydrogen Permeation Testing: Electrochemical vs. Pressurized Methods, Phillip Rückeshäuser [phillip.rueckeshaeuser@tuwien.ac.at], TU Wien, Austria; *Szilard Kolozsvari, Peter Polcik,* Plansee Composite Materials GmbH, Germany; *Timea Stelzig,* Oerlikon AM Europe GmbH, Germany; *Konrad Fadenberger,* Oerlikon Balzers Coating Germany GmbH, Germany; *Klaus Boebel,* Oerlikon Balzers, Liechtenstein; *Tomasz Wojcik, Helmut Riedl,* TU Wien, Austria

The interactions between hydrogen and materials are subjects of significant interest in both research and industry. Consequently, the hydrogen charging of samples and following detection are crucial aspects of this field. Currently, there are two primary methods for hydrogen charging: electrochemical charging and pressurized hydrogen charging.

In electrochemical charging, atomic hydrogen is produced through the dissociation of water, while in pressurized charging, hydrogen is introduced in gaseous form and subsequently thermally dissociated. Although these two methods are based on fundamentally different principles, they ultimately achieve the same goal: the absorption and permeation of atomic hydrogen in the sample. However, most studies tend to focus on either one method or the other. Thus, it becomes essential to explore the correlation between these methods and identify equivalent charging conditions for specific hydrogen permeation characteristics.

In this study, we conducted hydrogen permeation tests using both electrochemical and pressurized setups. We applied nitride coatings such as TiN and CrN to steel substrates using physical vapor deposition (PVD) techniques and compared the permeation performance of the two charging methods. This comparison involved determining key parameters such as diffusion coefficients and permeation reduction factors. Additionally, we characterized the coating properties using secondary electron microscopy, X-ray diffraction, and linear sweep voltammetry.

TS1-ThP-2 Towards Defect-Free Laser-Induced Graphene Coating on Copper and Aluminum Foils for Anode-Free Li and Na Metal Batteries, Aarti Gunjal [aartitambe22@gmail.com], IISER PUNE, India; *Suparna Saha,* TCG-CREST Kolkata, India; *Swati Jadhav,* IISER PUNE, India; *Satishchandra Ogale,* TCG-CREST Kolkata, India

Laser-induced graphene (LIG) is a very well-established process for realizing functional carbon coatings on different substrates. Depending on the laser processing parameters it yields structurally, morphologically and chemically interesting forms which are also adherent because of the thermal energy input. In most cases, however, a defect band (D band) is invariably observed in the Raman spectrum of such coatings, in addition to the so-called G-band corresponding to desired graphitic carbon. In this work we have applied CO₂ laser processing to carbonize the coatings of the dried powder of aromatic eucalyptus (EU) leaves on metals. The effects of laser power density, scanning speed, substrate choice, and precursor coating thickness on the quality of carbon are carefully examined and optimized to obtain a unique almost defect-free few layers graphene coating, as reflected by a miniscule D-band and a significant 2D signature in the Raman spectra of few-layer graphene. Comparative studies on other aromatic leaves sample (e.g. lemon grass) and non-aromatic leaves suggest that aromaticity is important for realizing an enhanced 2D band, which could be due to internal turbulent cavitation upon laser-induced transient heating. In situ nitrogen doping is also achieved by applying the LIG process to a mixture of EU powder loaded with urea. Remarkably, the nitrogen incorporated coatings on copper and aluminum current collectors render far superior performance in anode-free Li and Na metal batteries, respectively, as against the defect-free few layers graphene coating. In fact, half cells with alkali metals show an impressive cycling stability of over 450 cycles for both Li and Na. Through thorough characterizations employing multiple techniques, an attempt is made to develop a mechanistic understanding of the issues involved.

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