

## Protective and High-temperature Coatings Room Town & Country A - Session MA2-1-TuA

### Thermal and Environmental Barrier Coatings I

**Moderators:** Sabine Faulhaber, University of California, San Diego, USA, Fernando Pedraza, La Rochelle University, Laboratory LaSIE, France, Francisco Javier Perez Trujillo, Universidad Complutense de Madrid, Spain, Gustavo García-Martín, REP-Energy Solutions, Spain

4:00pm **MA2-1-TuA-8 Multicomponent Rare Earth Oxide Coatings for Refractory Alloys**, Rachel Rosner, Kristyn Ardrey, Will Riffe, Alejandro Salanova, Prasanna Balachandran, Bi-Cheng Zhou, Carolina Tallon, Jonathan Laurer, Jon Ihlefeld, Patrick Hopkins, Sandamal Witharamage, Elizabeth Opila [opila@virginia.edu], University of Virginia, USA **INVITED**

Rare earth oxide ( $RE_2O_3$ ) exhibit three crystal structures across the lanthanide series: hexagonal, monoclinic, and cubic, with all showing exceptionally high-melting temperatures ( $>2100^\circ C$ ) and excellent thermochemical stability. The cubic  $RE_2O_3$ , dysprosium through lutetium oxides, have isotropic thermal expansion with a reasonable match to Nb, making them suitable high temperature coatings for oxidation-prone refractory alloys. Multicomponent rare-earth oxides (MRO) allow the additional ability to target and optimize thermal expansion, resistance to molten deposits, and especially thermal conductivity, enabling their use as thermal/environmental barrier coatings (T/EBCs) in high-temperature, reactive environments such as turbine engines. Thermal conductivity of MROs has been shown to decrease with mixtures of  $RE_2O_3$  with increasing mass and size variation. The larger, lighter, non-cubic lanthanide oxides, lanthanum through terbium oxides, mixed in a majority MRO cubic phase in non-equimolar proportions will precipitate as second phases once their solubility limit in the cubic  $RE_2O_3$  is exceeded, enabling further reductions in thermal conductivity. In this work, MRO compositions are systematically varied to aid in achieving targeted thermal conductivity, thermal expansion, and resistance to molten deposits. Powder mixtures were combined, ball milled, and sintered via spark plasma sintering. Room temperature thermal conductivity was measured using the laser-based time domain thermoreflectance method. Thermal expansion was determined by dilatometry or lattice parameter measurements as a function of temperature. Resistance to molten  $CaO-MgO-Al_2O_3-SiO_2$  was quantified after exposure at temperatures of  $1300-1500^\circ C$  for times between 1 and 96h. Here we evaluate whether a single layer MRO will meet all design requirements for a (T/EBC) enabling cost efficient coating synthesis or whether additional coating layers are required to achieve adherent, protective properties for hot section turbine engine component applications.

4:40pm **MA2-1-TuA-10 PVD Ce-Coating to Mitigate Intergranular Oxidation of Additively Manufactured Ni-Base Alloy In625**, Anton Chyrkin [chyrkin@chalmers.se], Andrea Fazi, Mohammad Sattari, Mattias Thuvander, Chalmers University of Technology, Gothenburg, Sweden; Wojciech J. Nowak, Rzeszow University of Technology, Rzeszow, Poland; Dmitry Naumenko, Forschungszentrum Jülich GmbH, Germany; Jan Froitzheim, Chalmers University of Technology, Gothenburg, Sweden

Additively manufactured (AM) Powder Bed Fusion - Laser Beam Ni-base alloy IN625 suffers from intergranular oxidation (IGO) during air exposure at  $900^\circ C$  in contrast to the conventionally manufactured (CM) forged alloy IN625. A new mechanism of IGO in AM alloys is proposed: IGO is triggered by oxide buckling over the grain boundaries (GBs) in the alloy followed by oxidation of the open intergranular voids. Application of a 10 nm thick PVD coating of Ce promoted a slower inward growth of the  $Cr_2O_3$  scale, better oxide adherence and as a result strongly suppressed IGO. The  $Cr_2O_3$  scales thermally grown on both uncoated and Ce-coated alloys were analysed with SEM/EDX, EBSD, GD-OES, TEM and APT. The main beneficial effect provided by the Ce-coating is improved oxide adherence preventing oxide from buckling and oxidation of intergranular voids.

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