

Atomic Scale Processing Mini-Symposium Room Ballroom A - Session AP+EL+PS+TF-ThP

Atomic Scale Processing Mini-Symposium Poster Session

AP+EL+PS+TF-ThP-1 Design of Gas Flow Field for a Slip Flow ALD Processing Chamber, Kyung-Hoon Yoo, Korea Institute of Industrial Technology, Republic of Korea; *Geun-Soo Song*, The KUMYOUNG Inc., Republic of Korea; *Chun-Sik Kim*, TNG Co., Republic of Korea; *Jun-Young Hwang*, *Sang-Ho Lee*, *Jeong-Jin Kang*, *Shin-Ae Song*, *Kyung-Tae Kang*, Korea Institute of Industrial Technology, Republic of Korea; *Hyeong-Cheol Lee*, Hanyang University, Republic of Korea; *Kun-Hyung Lee*, SAMSUNG DISPLAY, Republic of Korea

As semiconductor feature sizes reach the nanometer scale, ALD processing equipment has become indispensable for atomic-layer control. However, conventional ALD processes face significant challenges regarding economic and environmental feasibility due to the excessive consumption of precursors and energy.^{1,2} To address these issues, it is essential to establish sustainable manufacturing technologies through the development of high-efficiency ALD processing chambers. In the present study, we propose a slipflow based ALD chamber designed to optimize the process space and minimize resource waste. The nitrogen flow fields within the chamber were numerically investigated using Computational Fluid Dynamics (CFD) for process gap sizes of 1, 10, and 100 mm at 400 °C. Under an inlet static pressure of 1 Torr and a mass flow rate of 4.233×10^{-5} kg/s, the Knudsen number (Kn) and flow Reynolds number (Re) were evaluated as 0.0137 and 0.822, respectively for the 10 mm gap. These values confirm that the flow resides within the slip flow regime, where rarefaction effects at the surface wall become significant. Accordingly, the steady-state compressible laminar flow field was simulated by solving the continuity, momentum, and energy equations.^{3,4}

Acknowledgment

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AP+EL+PS+TF-ThP-2 Nucleation and Evolution of Dislocations and Extended Faults in Complex Perovskite Oxides, Rishi Raj, K. Andre Mkhoyan, University of Minnesota, USA

The stoichiometric flexibility of a perovskite oxide (ABO₃) results in several unique properties due to the presence of three elements in the crystal structure. This allows it to host a variety of defects across all dimensions. 1D primitive edge dislocations with varying terminations, screw dislocations as well as 2D structural faults like Ruddlesden-Popper, elastic distortions, and BO₃ octahedral tilts are commonly observed and studied. Even though structure and electronic properties of some of these defects have been reported, the nucleation and growth of such defects and dislocations is still unclear. Analytical STEM study of perovskite oxide thin films of BaSnO₃ and SrSnO₃ show the existence of two kinds of edge dislocations: single ([001]/(100)-type) and dissociated ([001]/(110)-type) edge dislocation each with two half planes missing. Six different types of such dislocations are possible and observed due to the terminations of the perovskite half planes. Similarly, 2D Ruddlesden-Popper faults are also observed to result from missing planes of atoms, suggesting a correlation with 1D dislocations. Apart from such conventional topological defects, complex perovskite oxides also showcase several line defects and dislocations with reconstructed cores. This suggests we can selectively populate the material with defects by altering the terminations of crystal half planes. Strain tuning during growth by swapping substrates or decorating the surface morphology can effectively allow termination control. This would allow certain defect and dislocation structures to be more favorable than others. By establishing this correlation between 1D and 2D faults, this work

Thursday Evening, November 12, 2026

provides a potential pathway to transition from passive extended defect management to intentional, termination-controlled defect engineering.

AP+EL+PS+TF-ThP-3 Atomistic Study of the Effect of Moisture-Modified Amorphous SiO₂ Interfaces on Gas-Phase Clustering in LPCVD, Minseung Cha, Seoul National University, Republic of Korea

Particle formation is a critical issue in LPCVD processes because nanoscale contaminants can cause film defects, pattern failure, yield loss, and process instability. Although particle generation is often explained by precursor decomposition and homogeneous gas-phase reactions, the chemical state of quartz-glass components used in LPCVD equipment has not been sufficiently considered. In particular, amorphous SiO₂ surfaces are widely exposed in parts such as process tubes, chambers, boats, injectors, and nozzles, and these surfaces can be modified by moisture during air exposure.

In this study, we investigate the role of moisture-modified amorphous SiO₂ interfaces in gas-phase clustering using molecular dynamics and density functional theory calculations. Dry SiO₂, hydroxylated SiO₂ surface models are constructed to represent different surface states. H₂O adsorption and dissociative chemisorption are first examined to evaluate silanol formation. The interaction between reactive molecules or cluster precursor species and each SiO₂ surface is then analyzed in terms of adsorption stability, residence behavior, and desorption tendency.

We propose that hydroxylated SiO₂ surfaces can act as reactive boundaries rather than inert walls in LPCVD environments. Surface silanol groups may temporarily stabilize reactive species through hydrogen bonding and electrostatic interactions, increasing local molecular concentration near the interface. These stabilized species can subsequently return to the gas phase and enhance molecular aggregation, thereby increasing the probability of cluster and particle formation.

This study provides an atomistic interpretation of how the moisture exposure condition of quartz-glass parts can influence gas-phase clustering in LPCVD processes. The results suggest that controlling moisture exposure and the hydroxylation state of amorphous SiO₂ interfaces in components such as nozzles, chambers, and boats may be important for particle suppression and process stability.

Keywords: LPCVD, amorphous SiO₂, H₂O vapor, hydroxylation, silanol, molecular dynamics, MD, density functional theory, DFT, gas-phase clustering, particle formation, quartz glass, semiconductor, process

AP+EL+PS+TF-ThP-4 Low Temperature Atomic Layer Deposition of ZnO for Optoelectronic Applications, Anil Yadav, Aditya Sadhanala, Indian Institute of Science, India

This work demonstrates the ultra-low-temperature atomic layer deposition (ALD) of ZnO thin films at 50 °C and 70 °C using a 1 M diethylzinc (DEZ) solution in hexane and water as precursors. This was achieved in a custom-built ALD chamber with precise control over the process parameters. Despite the ultra-low deposition temperatures, the resulting ZnO films demonstrated excellent uniformity. This approach not only solves the problem of substrate compatibility by enabling ultra-low-temperature growth but also demonstrates the use of solvent-diluted pure precursors. The X-ray diffraction (XRD) patterns of as-deposited ZnO films deposited at 50 °C and 70 °C as a function of the number of ALD cycles from 50 to 250 were measured. As the ALD deposition cycles increased to 100, distinct diffraction peaks corresponding to (100), (002), and (101) planes emerged. When the deposition temperature increased from 50 °C to 70 °C, the (002) peak became more intense and sharper, indicating improved crystallinity.

To evaluate the electronic properties of ZnO thin films deposited at ultra-low temperatures of 50 °C and 70 °C, we integrated them as the active layer in field-effect transistors (FETs). The fabricated devices demonstrated exceptional electrical characteristics, exhibiting an I_{ON}/I_{OFF} ratio exceeding 10⁸ with the saturation mobility of 0.85 cm²/V.s. These results clearly confirm the applicability of solution-based precursors for ultra-low-temperature ALD depositions in electronic device applications.

AP+EL+PS+TF-ThP-5 Oxidation-Assisted SF₆ Remote Plasma Etching for Surface Roughness Reduction of TiN, Min Kyun Sohn, Subin Heo, Seong Hyun Lee, Sun Kyu Jung, Jeong Woo Park, Dongwoo Suh, Electronics and Telecommunications Research Institute, Republic of Korea

Titanium nitride (TiN) is widely used as a conductive diffusion barrier and work-function metal in advanced semiconductor devices. As device dimensions continue to scale down, controlling not only the etch amount but also the post-etch surface morphology becomes increasingly important, because nanoscale roughness can directly affect electrical reliability,

interface quality, and subsequent film deposition. In the present study, we investigate the effect of SF₆ remote plasma-based etching on the surface morphology of TiN thin films, with particular emphasis on the role of a pre-oxidation step.

Three TiN surface conditions were compared: as-deposited TiN, TiN exposed to SF₆ radicals only, and oxidized TiN followed by SF₆ radical exposure. The etch per cycle (EPC) increased from 0.16 Å/cycle for the SF₆-only process to 0.38 Å/cycle for the oxidation-assisted SF₆ radical process, indicating that surface oxidation enhanced the fluorine-radical-driven removal of TiN. Atomic force microscopy measurements showed that the as-deposited TiN exhibited an RMS roughness of 1.313 nm, which modestly decreased to 1.117 nm after direct SF₆ radical exposure. In contrast, when the TiN surface was oxidized prior to SF₆ radical exposure, the RMS roughness further decreased to 0.850 nm, corresponding to an approximately 35% reduction compared with the as-deposited TiN surface, demonstrating a clear surface-smoothing behavior.

This difference suggests that the oxidation step plays a critical role in modifying the TiN surface before fluorine-radical exposure. The smoothing behavior is attributed to the formation of a TiO_x or TiO_xN_y surface layer during oxidation, followed by its conversion into volatile TiF_x species during SF₆ radical exposure. The higher EPC and lower RMS roughness obtained after oxidation-assisted SF₆ exposure suggest that the modified surface layer can be removed more efficiently and uniformly than the native TiN surface. This oxidation-assisted pathway provides a chemically controlled route for reducing TiN surface roughness without relying on ion-driven physical sputtering.

These results indicate that surface pre-oxidation can be an effective strategy for controlling the morphology of TiN during radical-based etching. The proposed oxidation-assisted SF₆ radical process offers a potential strategy for morphology-controlled TiN etching in nanoscale device integration.

AP+EL+PS+TF-ThP-6 Development of in-Situ Laser Diagnostics Measurements of CH₃ Radical Concentrations for Study of Thin Film Depositions Using Organo-Metallic Precursors, Mruthunjaya Uddi, Advanced Cooling Technologies Inc.; *Prawal Agarwal, Devon Jensen, ACT; Anatoli Morozov, Princeton University; Arthur Dogariu, Texas A&M University*

Many environmentally friendly non-halogen precursors contain CH₃ structure component. Plasma chemical vapor deposition of pure metallic thin films using these precursors is challenging and requires better understanding of the gas phase and surface reaction mechanisms. We present a novel, time and spatially resolved, in-situ femtosecond laser diagnostics method to measure CH₃ radicals in a non-equilibrium plasma environment. The CH₃ radicals are predissociated to CH₂ and H radicals using a 275 nm femtosecond laser pulse, followed by a 205 nm probe beam to measure the H radical concentrations. A single femtosecond laser is used to generate the required 275 nm and 205 nm beams. Measuring and knowing the dissociation cross section of CH₃ radical by 275 nm pump beam, allows the quantification of CH₃ radical concentrations with spatial and time resolution.

AP+EL+PS+TF-ThP-7 Hybrid Electrospun ZnO Nanofiber and ALD ZnO Electron Transport Layers for Near-Infrared Organic Photodiodes, Anna Leonard, *Neha Chaturvedi, Veena Misra, North Carolina State University*

Organic photodiodes (OPDs) require efficient electron transport layers (ETLs) to achieve charge-selective extraction while minimizing recombination and dark current impacts. In this work, hybrid ZnO ETLs consisting of electrospun ZnO nanofibers and atomic layer deposited (ALD) ZnO were investigated for integration into near-infrared OPDs with a PTB7-Th:IEICO-4F active layer. The inclusion of the electrospun ZnO nanofibers increases the interfacial area between the ETL and active layer and provides enhanced electron transport pathways while the ALD ZnO layer provides film continuity and assists in passivating traps to mitigate defect states introduced by the nanofibers.

Three ETL architectures were investigated: ZnO fibers beneath ALD ZnO, ZnO fibers above ALD ZnO, and ALD ZnO control devices. ZnO nanofibers were fabricated through electrospinning of a zinc acetate dihydrate/polyvinylpyrrolidone (PVP) solution followed by calcination to oxidize the fibers and remove excess PVP. Device performance was evaluated through dark current density vs. voltage (J_{dark} -V) and external quantum efficiency (EQE) measurements. Devices utilizing hybrid nanofiber/ALD ZnO ETLs exhibited reduced initial dark current compared to control devices indicating suppression of leakage pathways and trap-assisted recombination. Enhanced EQE response was observed in devices

Thursday Evening, November 12, 2026

with fibers deposited above the ALD ZnO layer, suggesting improved interface quality and charge extraction.

These results demonstrate that combining electrospun ZnO nanostructures with ALD ZnO provides a promising method to engineer ETLs for high-performance OPDs and other optoelectronic devices.

AP+EL+PS+TF-ThP-8 Site-Specific and Temperature-Dependent Hydration of Faceted Hematite, Asare Dua, *Illinois Institute of Technology; Luke Pretzie, Purdue University; Ashley Bielinski, Argonne National Laboratory; Yue Li, Argonne National Lab; Valentine Novosad, Cong Liu, Alex Martinson, Argonne National Laboratory; Adam Hock, Illinois Institute of Technology*

Alpha hematite (α -Fe₂O₃) is an abundant metal oxide whose surfaces and interfaces control key processes in catalysis, sensing, and photoelectrochemistry. While the more stable α -Fe₂O₃ (001) surface is well studied, less stable facets such as (012) and (104) which are more relevant in the aforementioned applications due to higher surface activity remain least studied. Surface sites of α -Fe₂O₃ (012) and (104), as well as their distinct stability were identified through temperature-dependent hydration by combining in situ infrared reflection absorption spectroscopy (IRAS) with density functional theory (DFT) vibrational analysis. For α -Fe₂O₃ (012), we found sites that promoted the dissociative and molecular adsorption of D₂O. Dissociatively adsorbed D₂O were either terminal or bridging. In both cases the hydroxyls were either isolated or interacting with nearby species. Protons of bridging hydroxyls (μ 3a-OD (isolated); μ 3b-OD (interacting with nearby hydroxyls)) are bound to triply coordinated surface oxygens while terminal hydroxyls (μ 1a-OD (interacting with nearby hydroxyls)) were bound to octahedral surface Fe atoms (Fe_{oct}³⁺). Similar to dissociatively adsorbed D₂O, molecularly adsorbed D₂O was either isolated (D₂Oa) or interacting with nearby hydroxyls (D₂Ob). α -Fe₂O₃ (104) on the other hand exhibited an isolated doubly coordinated bridging hydroxyl (μ 2a-OD) and an interacting bridging species (μ 2b-OD) rather than triply coordinated bridging hydroxyls. This preference is backed by our DFT calculations which shows triply coordinated bridging hydroxyls were highly unstable on α -Fe₂O₃ (104). Results from IRAS showed two isolated terminal hydroxyls, μ 1a-OD and μ 1b-OD, were present on the surface of α -Fe₂O₃ (104). This indicates two types of undercoordinated surface atoms existed with the first being undercoordinated tetrahedral Fe atoms (Fe_{tet}³⁺) since they're the surface atoms of α -Fe₂O₃ (104), while the second is from the second layer of Fe atoms which have octahedral geometry. The undercoordination of Fe_{oct}³⁺ atoms however suggests oxygen vacancies (V_o) were present and some of these might migrate from surface to the second lattice of α -Fe₂O₃ (104). Our temperature dependent studies support this hypothesis since the population of μ 1a-OD, μ 1b-OD, and μ 2a-OD increased with increasing temperature when under vacuum (0.8 torr). The distinct local environments of the sites on α -Fe₂O₃ (012) and α -Fe₂O₃ (104) and changes in properties with respect to temperature revealed through this work provide a fundamental tool that may be used to engineer the surface of alpha hematite through site- or facet-selective atomic layer deposition.

Author Index

Bold page numbers indicate presenter

— A —

Agarwal, Prawal: AP+EL+PS+TF-ThP-6, **2**

— B —

Bielinski, Ashley: AP+EL+PS+TF-ThP-8, **2**

— C —

Cha, Minseung: AP+EL+PS+TF-ThP-3, **1**

Chaturvedi, Neha: AP+EL+PS+TF-ThP-7, **2**

— D —

Dogariu, Arthur: AP+EL+PS+TF-ThP-6, **2**

Dua, Asare: AP+EL+PS+TF-ThP-8, **2**

— H —

Heo, Subin: AP+EL+PS+TF-ThP-5, **1**

Hock, Adam: AP+EL+PS+TF-ThP-8, **2**

Hwang, Jun-Young: AP+EL+PS+TF-ThP-1, **1**

— J —

Jensen, Devon: AP+EL+PS+TF-ThP-6, **2**

Jung, Sun Kyu: AP+EL+PS+TF-ThP-5, **1**

— K —

Kang, Jeong-Jin: AP+EL+PS+TF-ThP-1, **1**

Kang, Kyung-Tae: AP+EL+PS+TF-ThP-1, **1**

Kim, Chun-Sik: AP+EL+PS+TF-ThP-1, **1**

— L —

Lee, Hyeong-Cheol: AP+EL+PS+TF-ThP-1, **1**

Lee, Kun-Hyung: AP+EL+PS+TF-ThP-1, **1**

Lee, Sang-Ho: AP+EL+PS+TF-ThP-1, **1**

Lee, Seong Hyun: AP+EL+PS+TF-ThP-5, **1**

Leonard, Anna: AP+EL+PS+TF-ThP-7, **2**

Li, Yue: AP+EL+PS+TF-ThP-8, **2**

Liu, Cong: AP+EL+PS+TF-ThP-8, **2**

— M —

Martinson, Alex: AP+EL+PS+TF-ThP-8, **2**

Misra, Veena: AP+EL+PS+TF-ThP-7, **2**

Mkhoyan, K. Andre: AP+EL+PS+TF-ThP-2, **1**

Morozov, Anatoli: AP+EL+PS+TF-ThP-6, **2**

— N —

Novosad, Valentine: AP+EL+PS+TF-ThP-8, **2**

— P —

Park, Jeong Woo: AP+EL+PS+TF-ThP-5, **1**

Pretzie, Luke: AP+EL+PS+TF-ThP-8, **2**

— R —

Raj, Rishi: AP+EL+PS+TF-ThP-2, **1**

— S —

Sadhanala, Aditya: AP+EL+PS+TF-ThP-4, **1**

Sohn, Min Kyun: AP+EL+PS+TF-ThP-5, **1**

Song, Geun-Soo: AP+EL+PS+TF-ThP-1, **1**

Song, Shin-Ae: AP+EL+PS+TF-ThP-1, **1**

Suh, Dongwoo: AP+EL+PS+TF-ThP-5, **1**

— U —

Uddi, Mruthunjaya: AP+EL+PS+TF-ThP-6, **2**

— Y —

Yadav, Anil: AP+EL+PS+TF-ThP-4, **1**

Yoo, Kyung-Hoon: AP+EL+PS+TF-ThP-1, **1**