

Fig 1. Thermogravimetric analysis of $(t\text{BuN})_2\text{MoCl}_2 \cdot (t\text{BuDAD}^{\text{H}})$ (**1**, red) and $(t\text{BuN})_2\text{MoCl}_2 \cdot \text{bpy}$ (**2**, blue) with heating rates of $10\text{ }^\circ\text{C min}^{-1}$. The mass loading of each sample was 10.0 mg.

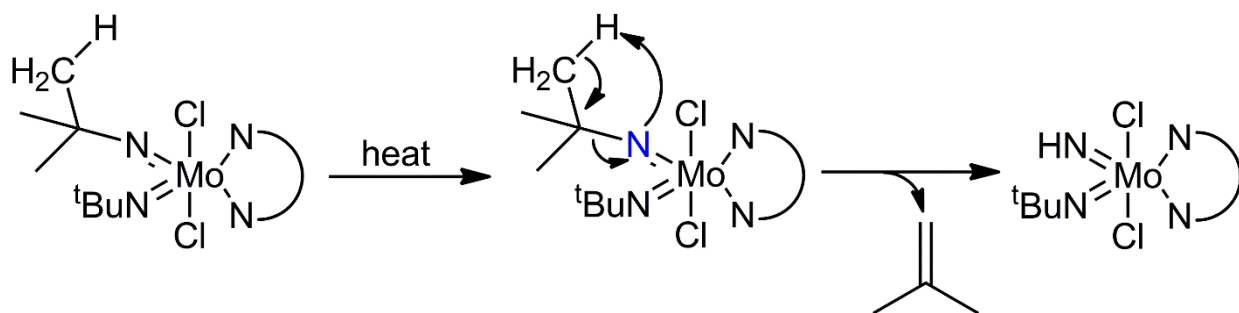


Fig 2. Pictorial representation of the proposed thermal decomposition of the $(t\text{BuN})_2\text{MoCl}_2$ adducts, *via* γ -H activation of the *tert*-butylimido group.

- [1] Land, M. A.; Robertson, K. N.; Barry, S. T. *Organometallics*, **2020**, *39*, 916–927. [doi.org/acs.organomet.9b00578](https://doi.org/10.1021/acs.organomet.9b00578).